UNIVERSITY OF DIYALA College of Medicine Dept. of Chemistry and Biochemistry



SYLLABUS OF

MEDICAL CHEMISTRY AND BIOCHEMISTRY

THEORY

1 st stage



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Vision, message and objectives of the branch:

Vision:

- 1. Actively contribute to medical progress through education and prepare qualified doctors to provide the best medical services and continue scientific research in all medical fields.
- 2. The preparation of doctors with competence and scientific experience enhanced by understanding the biochemical foundations of vital processes that occur within the human body in normal and pathological cases.

Message:

- 1. Excellence in the innovation and follow-up of advanced scientific methods in the conduct of pathological analyses and the preparation of medical scientific research that contributes to the service of the community.
- 2. Establishing sober relationships with researchers at international universities.

Goals:

- 1. Keeping up with the scientific development in the development of education programs and using the latest programs developed for medical education in accordance with the modern academic curriculum.
- 2. Contribute to providing the community with scientifically distinguished doctors with experience in the scientific foundations adopted to conduct all clinical biochemistry analyses.
- 3. Lectures for graduate students at university colleges as well as supervising their research projects.

<u>Credit hours</u> :

Theory hours of the year : 90

Theory hours of the course : 45

Practical hours of the year : 60

Practical hours of the course : 30

Theory Exam:

Mid year exam : is a 2 hrs.

Final year exam : is a 3 hrs.

Practical Exam :

Mid year exam : is a 2 hr.

Final year exam : is a 3 hrs.

Learning units :

Units of teaching of the year : 8

Units of teaching of the course / Theoey : 6

Units of teaching of the course / Practical : 2

first

Semester

Total weeks of the course : 15

Theory hours of the course : 45

Theory hours / week : 3

Units of teaching: 3

Fluid and Electrolyte Balance

- 1. Define body fluid and electrolytes.
- 2. Know the volumes and main composition of body fluids.
- 3. List the factors that determine body water content and describe the effect of each factor.
- 4. Describe the role of the body systems in regulating the body's fluid composition and volume.
- 5. Describe mechanisms that regulate water intake and hormonal controls of water output in urine.

Acid-Base Balance

- 1. Defines acids, bases.
- 2. Know the natural acids and bases ratio of the body.
- 3. Recognize the types of acid and base.
- 4. List the source of acids and bases of the body.
- 5. Study the systems responsible for maintenance of the acid-base balance.
- 6. Explain the role of buffer systems in regulating the pH of the intracellular fluid and the extracellular fluid.
- 7. Discuss acid base disorders
- 8. Analysis of Acid-Base Imbalances Report

Chemistry of Carbohydrates-1 Monosaccharides & Disaccharides

- 1. Define carbohydrate and the groups of saccharides
- 2. Know the chemical structure of the common sugars.
- 3. Understand the concepts of and isomerism in simple sugars anomers.
- 4. Glycosides, sugar alcohols, sugar acids, phosphate esters, deoxy sugars and amino sugars.
- 5. Understand the role saccharides play in biology
- 6. Know the biochemical functions and differences between the various heteropolysaccharides
- 7. Be able to recognize the N and O linked polysaccharides
- 8. Know how dietary polysaccharides are digested by humans

(9 hours)

Chemistry of Carbohydrates-Polysaccharides Part-2

- 1. Study the chemical structure of polysaccharides
- 2. Classify polysaccharides
- 3. Know the biochemical functions and differences between the various heteropolysaccharides
- 4. Be able to recognize the N and O linked polysaccharides
- 5. Know how dietary polysaccharides are digested by humans

Fatty acids & Derivatives

- 1. Have general idea about lipid structure and properties
- 2. Classify lipids
- 3. List the major physiological functions of fatty acids
- 4. Derive the structure of saturated or unsaturated fatty acids.
- 5. Study the relation between the structure and function of fatty acids
- 6. Be able to specify the omega or delta ends. Recognize the alpha, beta and gamma carbons of fatty acids
- 7. List and be able to identify the general features of the ecosanoids.
- 8. Know the biochemical functions of the eicosanoids

Glyceride, Non-glyceride & Complex lipids

- 1. Classify lipids.
- 2. Know the mean class of lipids
- 3. Have an idea about the structure of each class.
- 4. Understand the physical and chemical of the classes.
- 5. List the biological function of all classes.
- 6. Relate the structure and properties with the diseases come as a result of this lipids.

second Semester

Total weeks of the course : 15

Theory hours of the course : 45

Theory hours / week : 3

Units of teaching: 3

Amino Acids & Proteins Part 1

- 1. Describe the general structure of an amino acid.
- 2. Recognize amino acids and classify them based on the characteristics of their side chains.
- 3. List the twenty common amino acids found in living organisms.
- 4. Describe how a peptide bond forms.
- 5. Understand the biologic activities of peptides

Amino Acids & Proteins Part 2

- 1. Understand that amino acids are linked via peptide bonds to make polypeptides and proteins
- 2. Understand that each protein molecule can be hundreds of amino acids long and the amino acids must be joined in a precise order.
- 3. Know that the side-chains (R groups) of the amino acids can interact with one another to fold the protein into a particular shape which is essential for the protein to function correctly.
- 4. Describe, using examples, the relationship between protein structure and function.
- 5. Define denaturation and list factors led to protein denaturation
- 6. List some medical application of denaturation

(**3 hours**)

Amino Acids & Proteins Part 3

- 1. Classify proteins according to different parameters including chemical composition, shape, biological function, solubility in water.
- 2. Describe, using examples, the relationship between protein structure and function
- 3. Explain of biological activity of some important proteins

Fourth Subject

(**6 hours**)

Nucleic Acids Part 1

- 1. Describe the structure of a nucleotide as being a phosphate group, pentose sugar (either ribose or deoxyribose), and a nitrogen containing base,
- 2. Recall that the nitrogenous bases are adenine, cytosine, guanine, and thymine in DNA, or uracil in RNA, and the base pairings that occur,
- 3. State that a nucleic acid is formed from many nucleotides, joined by condensation reactions,
- 4. Compare and contrast the structures of DNA and RNA,
- 5. Explain the importance of DNA in storing genetic material and safely transferring genetic information between organisms.

Fifth Subject

(**6 hours**)

Nucleic Acids Part 2 Protein Biosynthesis

- 1. Comprehend the universal nature of the gene.
- 2. Be able to define replication of DNA.
- 3. Know the roles of mRNA, ribosomes, tRNA and amino acids in the process of translation.
- 4. Understand what start codons and stop codons are.
- 5. Understand how a polypeptide is built, one amino acid at a time, in the different docking sites of the ribosome.
- 6. Understand how tRNAs are 'charged' with amino acids.
- 7. Know that ribosomes consist of a large and a small subunit.
- 8. Be able to define polysome.

Sixth Subject

(**6 hours**)

Nucleic Acids Part 3

- 1. Define how errors by DNA polymerase create mutations
- 2. Identify the types of gene mutations.
- 3. Describe what occurs during each type of mutation.
- 4. Explain the structure and shape of viruses.
- 5. Know the viral replication, viral transaction and viral protein biosynthesis.
- 6. Discuss how to prevent viral transaction and viral protein biosynthesis

(**6 hours**)

Enzymes Part 1

- 1. Define enzyme and explain basic functions of enzymes
- 2. Explain basic properties of enzymes
- 3. Discover and defines the enzyme components
- 4. Express localization of enzymes in the cell
- 5. Defines the active site and catalytic activity of enzyme
- 6. Discuss working principle of enzymes
- 7. Express the relationship between enzyme and substrate

Eigth Subject

(**6 hours**)

Enzymes Part 2

- 1. Explain what an enzyme inhibitor is.
- 2. Distinguish between reversible and irreversible inhibitors.
- 3. Differentiate between competitive and noncompetitive inhibitors.
- 4. Discuss the biological role of isoenzymes and their use in clinical diagnosis.
- 5. Understand the bases of enzyme catalysis and the mechanisms of enzyme regulation.
- 6. Know the role of regulatory enzymes in controlling metabolic pathways and cellular responses.

Teaching and learning methods :

- 1. Method of givin lectures .
- 2. Student groups .
- 3. Workshops.
- 4. Reports & Research

Teaching Tools :

- 1. Screen LCD.
- 2. Data show .
- 3. For experimental part :
 - a. Spectrophotometer .
 - b. Centrifuge.
 - c. Water bath .
 - d. Incuvate .
 - e. Test tubes .
 - f. Syringes .

Students assessment methods:

- 1. Examination.
- 2. Reports preparation .

Questions include:

- 1-MCQs. $\sqrt{}$
- 2-Single choice questions. $\sqrt{}$
- 3-Matching. $\sqrt{}$
- 4-(Modified)Essay questions $\sqrt{}$
- 5-etc.....

Marks of theory , practical :

Total marks

- Theory : 75 %
- Practical : 25 %

First course Examination

- Theory : 15
- Practical : 5

Second course Examination

- Theory : 15
- practical : 5

Final Examination

- Theory : **45** Practical : **15**

Reference :

- Lippincott Illustrated Reviews : Biochemistry , Seventh Edition , 2018 .

- Harper's Biochemistry, 31 ST Edition, 2018.

- Lehninger Principle of Biochemistry, 4 th Edition, 2005.

- Essentials of Medical Biochemistry with clinical cases , 3 rd Edition , 2022 . By N.V. Bhagavan and chury – Eun Ha .

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UNIVERSITY OF DIYALA College of Medicine Dept.of Chemistry and Biochemistry



SYLLABUS OF

MEDICAL CHEMISTRY AND BIOCHEMISTRY

PRACTICAL

1 st stage

2023-2024

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- 2. Establishing sober relationships with researchers at international universities.

<u>Objectives</u> :

- 1. Keeping up with the scientific development in the development of education programs and using the latest programs developed for medical education in accordance with the modern academic curriculum.
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Units of teaching of the course / Theoey : 6

Units of teaching of the course / Practical : 2

First

Semester

Total weeks of the course : 15

Practical hours of the course : 30

Practical hours / week : 2

Units of teaching: 1

First Week

(**2 hours**)

Laboratory safety

- a. Understand the proper laboratory safety.
- b. Increase the awareness of the possible risks or hazards involved with laboratory work.
- c. Realize the laboratory is generally a safe place to work if safety guidelines are properly followed.

Laboratory instruments and apparatuses

- a. Identify and categorize the different instruments and apparatuses with their parts and uses in practice .
- b. Identify the photometer with its main parts and uses

Units and references values

- a. Recognize the principles of photometry and the related laws.
- b. Measure weight and volume

Fourth Week

(**2 hours**)

Applications of spectrophotometers

- a. Learn the purpose and proper use of a spectrophotometer.
- b. Determine the relationship between light absorbance and the number of particles in a sample in a given volume.
- c. Apply different methods for expressing concentration .
- d. Prepare stock solutions and perform different dilutions

Fifth Week

(**2 hours**)

Blood components

- a. Describe the blood components in details.
- b. Explain the blood samples in details.

Preparation of plasma and serum for analysis

- a. Describe the blood samples in details.
- b. Outline the importance of blood samples.

Sample collection, processing and handling

- a. Outline the type of biological samples .
- b. Describe the Blood collection techniques .

Eigth Week

(**2 hours**)

pH and Buffer, Acid- Base Balance

- a. Explain the acid base balance.
- b. Describe the role of buffers in maintaining the pH of a solution in body fluids.

Buffers in blood

- a. Identify the most powerful buffer systems in the body.
- b. Outline the importance of the buffer systems.

Urinalysis (UA)

- a. Outline the importance of urine samples
- b. Describe the collection of urine samples
- c. Describe urine examinations

Analysis of normal constituents of urine

- a. Describe the content of normal urine samples.
- b. Explain the results of urine examinations.

Analysis of abnormal constituents of urine

- a. Describe the content of abnormal urine samples.
- b. Explain the results of urine examinations for different cases.

General stool examination

- a. Outline the importance of stool samples
- b. Describe the collection of stool samples
- c. Describe stool examinations

Fourteenth Week

Hematological test

- a. Outline the importance of hematological test
- b. Explain the hematological test

Fifteenth Week

(**2 hours**)

First- semester examination

second Semester

Total weeks of the course : 15

Practical hours of the course : 30

Practical hours / week : 2

Units of teaching: 1

First Week

(**2 hours**)

Blood Glucose Test

- a. Identify the principles of the blood glucose test
- b. Calculation of glucose concentration in the unknown sample

Second Week

(**2 hours**)

Oral Glucose Tolerance Test

- a. Explain the types of the blood glucose tests
- b. Define the Oral Glucose Tolerance Test

Third Week

(**2 hours**)

Diabetes mellitus

- a. Describe diabetes mellitus.
- b. Explain its diagnosis and classification.

Fourth Week

(**2 hours**)

Case scenario of diabetes mellitus (Type I)

- a. Describe Type I diabetes mellitus.
- b. Illustration of case studies on Type I Diabetes Mellitus.

Fifth Week

(**2 hours**)

Case scenario of diabetes mellitus (Type II)

- a. a. Describe Type II diabetes mellitus.
- b. Illustration of case studies on Type II Diabetes Mellitus.

Sixth Week

(**2 hours**)

Lipid Profile

- a. Identify the principles of the lipid profile test
- b. Calculation of total cholesterol concentration in the unknown sample

Seventh Week

(**2 hours**)

Lipoproteins

- a. Identify the lipoproteins
- b. Estimate the concentration of HDL and LDL in the unknown sample

Eigth Week

(**2 hours**)

Plasma lipids and lipoproteins

- a. Describe disorders of lipid metabolism
- b. Illustration of case study

Ninth Week

(**2 hours**)

Case scenario of hypercholesterolemia

- a. Describe hypercholesterolemia.
- b. Illustration of case studies on hypercholesterolemia.

Tenth Week

(2 hours)

Case scenario of hypercholesterolemia in patients with diabetes mellitus

- a. Describe hypercholesterolemia in patients with diabetes mellitus .
- b. Illustration of case studies on hypercholesterolemia in patients with diabetes mellitus.

Eleventh Week

(**2 hours**)

Triglycerides

- a. Identify the principles of the triglycerides test
- b. Calculation of TG concentration in the unknown sample

Twelfth Week

(**2 hours**)

Case scenario of hyperlipidaemia

- a. Describe hyperlipidaemia .
- b. Illustration of case studies on hyperlipidaemia .

Thirteenth Week

(**2 hours**)

Case scenario of hyperlipidaemia in patients with diabetes mellitus

- a. Describe hyperlipidaemia in patients with diabetes mellitus .
- b. Illustration of case studies on hyperlipidaemia in patients with diabetes mellitus.

Fourteenth Week

(**2 hours**)

Case scenario of hypercholesterolemia and hyperlipidaemia in patients with diabetes mellitus

Objectives

a. Describe hypercholesterolemia and hyperlipidaemia in patients with diabetes mellitus .

b. Illustration of case studies on hypercholesterolemia and hyperlipidaemia in patients with diabetes mellitus .

Fifteenth Week

(**2 hours**)

Second-semester examination

Teaching and learning methods :

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